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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.003 Å R factor = 0.046 wR factor = 0.120 Data-to-parameter ratio = 12.2

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6-Benzylamino-2-(2-hydroxyethylamino)-9-methylpurine-1,7-diium bis(perchlorate) monohydrate

The title compound, $C_{15}H_{20}N_6O^{2+}\cdot 2ClO_4^{-}\cdot H_2O$, belongs to a group of cytokinin-derived compounds. It has been found that some cytokinin derivatives, e.g. 2,6,9-trisubstituted purine derivatives, behave as potent inhibitors of cyclin-dependent kinases (CDKs) and show anticancer activity. Olomoucine, i.e. the title cation in its neutral form, represents one of the first CDK specific inhibitors that selectively blocks CDK1, CDK2 and CDK5 kinases at micromolar concentrations. The asymmetric unit of the title compound consists of the diprotonated cation, two perchlorate anions and a water molecule of crystallization. The cation contains nearly planar benzene and purine ring systems, the dihedral angle between them being 46.77 $(6)^{\circ}$. The crystal structure is stabilized by $N_{purine} - H \cdot \cdot \cdot O_{water} / O_{hydroxy}$ $N_{amine} - H \cdot \cdot \cdot O_{water} / O_{hydroxy}$ $O_{water} - H \cdots O_{perchlorate}$ $O_{hydroxy} - H \cdots O_{perchlorate}$ and hydrogen bonds.

Comment

The frequent dysregulation of cancer cells has stimulated an intensive search for new chemical substances targeting cell division cycle events. The cell cycle is regulated by the timely and spatially coordinated action of cyclin-dependent kinases (CDKs), their positive and negative effectors. Inappropriate expression or mutations of CDKs, their modulators and even their substrates have often been detected in various cancers (Sherr, 1996). Therefore, CDK activity represents one of the most interesting targets for new generations of anticancer drugs. However, CDKs are also involved in other processes, such as apoptosis (CDK1, CDK5), neuronal functions (CDK5, CDK11) and transcription (CDK7, CDK8, CDK9). One can thus expect a large variety of cellular effects, and therefore also applications for CDK inhibitors. Besides oncology, the compounds are studied extensively for their promising influence on neurodegenerative disorders (Alzheimer's and Parkinson's diseases), cardiovascular diseases (restenosis), viral (HIV, Herpes, cytomegalovirus, papillomavirus) and parasitic (Plasmodium, Leishmania, Trypanosoma) infections and also for their potential use in in vitro reproduction and cloning (Meijer & Raymond, 2003).

The first compound identified as a specific and selective inhibitor of CDK1 and CDK2, 6-benzylamino-2-(2-hydroxyethylamino)-9-methylpurine, also named olomoucine (Veselý *et al.*, 1994), became a lead compound for the further design of improved drugs, *e.g.* roscovitine and purvalanols (Meijer & Raymond, 2003). The X-ray structure of, among others, the CDK2-olomoucine complex has also been determined (Schulze-Gahmen *et al.*, 1995). Many biological effects of CDK inhibitors in cellular assays have been described, and a significant correlation has been found between inhibition of Received 8 November 2004 Accepted 9 November 2004 Online 20 November 2004 CDKs and antiproliferative activity (Vermeulen *et al.*, 2002). In general, cell treatment with relevant concentrations of CDK inhibitors leads to dephosphorylation of the corresponding protein substrates and delay in progression through the cell cycle, to stimulation of *p*53-dependent transcription and subsequent synthesis of $p21^{WAF1}$ (Kotala *et al.*, 2001), to induction of apoptosis in cancer cells both *in vitro* and *in vivo* (McClue *et al.*, 2002).

Among known purine inhibitors, (R)-roscovitine (under synonym CYC202) is currently undergoing phase II clinical trials against lung and breast cancers and phase I tests against glomerulonephritis (Meijer & Raymond, 2003). In summary, the antiproliferative and proapoptotic effects of 2,6,9-trisubstituted purines suggest that these drugs have both an important and a promising anticancer potential.



The asymmetric unit of the title compound, (I), consists of a diprotonated organic cation, two perchlorate anions and a solvent water molecule (Fig. 1 and Table 1). It should be noted that the title cation represents only the fifth structurally characterized analog of 2,6,9-trisubstituted purines involving the 6-benzylaminopurine moiety (Cambridge Structural Database, Version 5.25.2; Allen, 2002) after (R)- and (S)-6-benzylamino-2-[(1-hydroxymethyl)propylamino]-9-isopropylpurine (Wang et al., 2001), 2,6-diamino- N^2 , N^6 -dibenzoyl-9-(α-L-threofuranosyl)purine (Wu et al., 2002; Schoning et al., 2002), N-[(2-azepan-1-yl)-9-isopropyl-9Hpurin-6-yl]-4-methoxybenzylamine (Trávníček & Zatloukal, 2004) and pentakis[trichloro(6-benzylamino-2-(3-hydroxypropyl)amino)-9-isopropylpurinium]platinum(II) (Trávníček et al., 2003). Moreover, the structure of the cation is also similar to those determined for 6-(4-methoxybenzylamino)purinium chloride (4MeOBapH; Trávníček et al., 2004), 6benzylaminopurinium bromide (BapH; Umadevi et al., 2001), 6-(2-chlorobenzylamino)purine (2ClBap; Maloň et al., 2001), 6-(3-chlorobenzylamino)purinium chloride (3ClBapH; Maloň et al., 2001), 6-(4-chlorobenzylamino)purinium perchlorate (4ClBapH; Maloň et al., 2002) and 6-(2-hydroxybenzylamino)purine (20HBap; Trávníček et al., 1997)

The cation contains nearly planar benzene (*A*), pyrimidine (*B*) and imidazole (*C*) ring systems, with maximum deviations from each plane of 0.006 (3), 0.013 (2) and 0.008 (2) Å for ring *A*, six-membered ring *B* and five-membered ring *C*, respectively (Nardelli, 1995). Atoms forming the purine ring (B + C)



Figure 1

The asymmetric unit of the title compound, showing the hydrogen bonding (dashed lines). Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres at arbitrary radii.

deviate slightly from planarity, the greatest deviation being 0.0149 (19) Å for atom N9. Planes *B* and *C* are nearly coplanar, with a dihedral angle of 0.79 (8)°, whilst the dihedral angles between planes *A* and *B*, and *A* and the purine ring (B + C) are 46.97 (7) and 46.77 (6)°, respectively. The $Cg1\cdots Cg2$, $Cg1\cdots Cg3$ and $Cg2\cdots Cg3$ distances are 6.3670 (2), 7.0460 (2) and 2.0817 (1) Å, respectively, where Cg1, Cg2 and Cg3 are the centroids of rings *A*, *B* and *C*, respectively. The torsion angles C6–N6–C9–C10, C9–N6–C6–C5 and N6–C9–C10–C15 are 154.8 (2), 171.0 (2) and 61.3 (3)°, respectively.

The cation is protonated at the N1 and N7 positions of the purine ring, *i.e.* it represents the N1-protonated N7 tautomer. It is evident that changes in protonation and substitution of the purine moiety cause changes in the interatomic parameters within the purine ring, mainly in the C-N-C angles. To date, 31 structures of compounds involving the 6-benzyl-aminopurine moiety have been deposited with the Cambridge Structural Database (Version 5.25.2; Allen, 2002). The main changes occur at the C2-N3-C4, C8-N7-C5 and C8-N9-C4 angles. A comparison of these interatomic parameters



Figure 2

Part of the crystal structure of the title compound, showing the hydrogen bonding (dashed lines). [Symmetry codes: (iii) 1-x, 1-y, 1-z; (iv) $1 - x, y - \frac{1}{2}, \frac{1}{2} - z.$]

for selected cytokinin-derivatives with the same group is given in Table 3.

The positive charge of the cation is compensated by two perchlorate anions. The crystal structure is stabilized by a network of N_{purine}-H···O_{water}/O_{hydroxy}, N_{amine}-H···O_{water}/ O_{hydroxy}, O_{water}-H···O_{perchlorate} and O_{hydroxy}-H···O_{perchlorate} hydrogen bonds, connecting adjacent cation, perchlorate anions and water molecules (Fig. 2 and Table 2).

Experimental

6-Benzylamino-2-(2-hydroxyethylamino)-9-methylpurine (olomoucine) was synthesized by a procedure similar to that described in the literature for the preparation of 2,6,9-trisubstututed purine derivatives (Imbach et al., 1999). Colorless crystals of the title compound suitable for single-crystal X-ray analysis were obtained by recrystallization of olomoucine from 2 M HClO₄. Elemental analysis (CHN Analyzer Flash EA 1112, ThermoFinnigen), calculated for

C15H22Cl2N6O10: C 34.83, H 4.29, N 16.25%; found: C 34.6, H 4.2, N 16.3%.

3569 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_o^2) + (0.06P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

 $R_{\rm int}=0.030$

 $\theta_{\rm max} = 25.0^{\circ}$

 $k=-8\rightarrow7$

 $l = -19 \rightarrow 19$

+ 4.28P]

 $(\Delta/\sigma)_{\rm max} = 0.001$ $\Delta \rho_{\rm max} = 0.94 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.63 \ {\rm e} \ {\rm \AA}^{-3}$

 $h = -21 \rightarrow 21$

Crystal data

$C_{15}H_{20}N_6O^{2+}\cdot 2ClO_4^{-}\cdot H_2O$	$D_x = 1.584 \text{ Mg m}^{-3}$
$M_r = 517.29$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 12391
a = 17.8770(7)Å	reflections
b = 7.4072 (3) Å	$\theta = 1.8 - 31.8^{\circ}$
c = 16.4214 (6) Å	$\mu = 0.37 \text{ mm}^{-1}$
$\beta = 94.187 \ (4)^{\circ}$	T = 100 (2) K
$V = 2168.69 (15) \text{ Å}^3$	Cube, colorless
Z = 4	$0.40 \times 0.40 \times 0.40$ mm

Data collection

Oxford Diffraction Xcalibur2 (Sapphire2 CCD) diffractometer ω scans Absorption correction: none 13724 measured reflections 3802 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.046$ $wR(F^2) = 0.120$ S = 1.003802 reflections 311 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1 Selected geometric parameters (Å, °).

N1-C6 1.366 (3) C5-N7 1.387 (3) N1-C2 1.394 (3) C5 - C61.406 (3) 1.329 (3) N2-C2N6-C61.317(3)C2-N31.327 (3) N6-C9 1.472 (3) N3-C4 1.336 (3) N7-C8 1.320 (3) C4-C5 1.378 (3) C8-N9 1.346 (3) C4 - N91.380 (3) C6-N1-C2 124.03 (19) C4-C5-C6 118.8 (2) N3-C2-N1 123.3 (2) N7-C5-C6 134.0(2)C2-N3-C4 112.54 (19) N1 - C6 - C5113.1(2)108.10 (19) N3-C4-C5 128.2 (2) C8-N7-C5 N3-C4-N9 125.2 (2) N7-C8-N9 110.0 (2) C5-C4-N9 106.61 (19) C8-N9-C4 108.05 (19) 107.25 (19) C4-C5-N7 C9-N6-C6-C5 171.0 (2) N6-C9-C10-C15 61.3 (3) C6-N6-C9-C10 154.8 (2)

Table 2	
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Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O1 - H1W \cdots O3^{i}$ $O1 - H1V \cdots O3^{ii}$	0.90 (3) 0.90 (3)	1.99 (3) 2.10 (3)	2.865 (3) 2.885 (3)	163 (4) 146 (4)
$O1 - H1V \cdots O4$	0.90 (3)	2.51 (3)	3.214 (3)	136 (3)
$N1 - H1 \cdots O10$ $N2 - H2 \cdots O10^{iii}$	0.88	2.18	2.794 (2) 2.942 (2)	134 145
$N6 - H6 \cdots O1$ $N7 - H7 \cdots O1$	0.88 0.88	2.17 1.86	3.024 (3) 2.688 (3)	162 156
$O10-H10\cdots O7^{iv}$	0.936 (10)	1.844 (16)	2.738 (3)	159 (3)

Symmetry codes: (i) x, y - 1, z; (ii) $2 - x, y - \frac{1}{2}, \frac{3}{2} - z$; (iii) 1 - x, 1 - y, 1 - z; (iv) $1-x, y-\frac{1}{2}, \frac{1}{2}-z.$

Table 3

Comparative bond angles (°) for selected cytokinin-derived compounds containing the 6-benzylaminopurine moiety.

Compound	C2-N3-C4	C8-N7-C5	C8-N9-C4
OlomoucineH ₂ ^a	112.54 (19)	108.10 (19)	108.05 (19)
NG38 ^b	110.77 (10)	103.50 (11)	105.95 (10)
4MeOBapH ^c	117.02 (16)	106.58 (16)	103.01 (15)
$BapH^{d}$	118.2 (7)	107.4 (6)	103.5 (6)
2ClBap ^e	111.32 (14)	103.68 (15)	106.19 (14)
3ClBapH ^f	117.6 (2)	106.8 (2)	102.60 (18)
4ClBapH ^g	113.8 (8)	109.1 (8)	103.7 (8)
	119.7 (8)	104.6 (8)	100.9 (8)
Bap^h	110.70	103.90	106.41
20HBap ⁱ	111.5 (3)	104.1 (3)	106.4 (3)
(R)-Roscovitine ^j	110.91 (17)	103.48 (18)	105.55 (17)
. /	111.56 (16)	103.58 (17)	105.39 (17)
$BTAP^k$	110.99 (26)	104.68 (26)	105.35 (24)

Notes: (a) this work, where olomoucineH₂ is the title cation; (b) Trávníček et al. (2004), where NG38 is N-[(2-azepan-1-yl)-9-isopropyl-9H-purin-6-yl]-4-methoxybenzylamine; (c) Trávníček et al. (2004), where 4MeOBapH is the 6-(4-methoxybenzylamino)purinium cation; (d) Umadevi et al. (2001), where BapH is the 6-benzylaminopurinium cation; (e) Maloň et al. (2001), where 2CIBap is 6-(2-chlorobenzylamino)purinium cation; (g) Maloň et al. (2001), where 3CIBapH is the 6-(3-chlorobenzylamino)purinium cation; (g) Maloň et al. (2002), where 4CIBapH is the 6-(4-chlorobenzylamino)purinium cation; (g) Maloň et al. (2002), where 4CIBapH is the 6-(4-chlorobenzylamino)purinium cation (the structure consists of two crystallographically independent molecules); (h) Raghunathan et al. (1983), where Bap is 6-benzylaminopurine; (i) Trávníček et al. (2004), where 2OHBap is 6-(2-hydroxybenzylamino)purine; (j) Wang et al. (2001), where (R)-roscovitine is (R)-6benzylamino-2-[(1-hydroxymethyl)propylamino]-9-isopropylpurine (the structure consists of two crystallographically independent molecules); (k) Wu et al. (2002), where BTAP is 2,6-diamino-N²,N⁶-dibenzoyl-9-(a-t-threofuranosyl)purine.

H atoms attached to C and N atoms were found in difference Fourier maps, idealized and refined using a riding model, with C–H distances of 0.95 and 0.99 Å and N–H distances of 0.88 Å, and with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm CH}, {\rm CH}_2{\rm and NH})$ or $1.5U_{\rm eq}({\rm CH}_3)$. All H atoms attached to O atoms were refined isotropically.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2004); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2004); data reduction: *CrysALis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP*III (Burnett & Johnson, 1996); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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